

Advanced Book Information

Rhenium Disulfide

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Handbook / color print, paperback

Rhenium disulfide, especially in low-dimensional form, is a subject of lively research into its electronic and optical properties. The field of twodimensional materials such as graphene and its analogues has been growing very rapidly.

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Materials Research Foundations Vol. 40 / **BISAC:** TEC021000 /**BIC/Thema:** TGM**Imprint:** Materials Research Forum LLC, *Publisher's sales rights are Worldwide***Summary:**

Rhenium disulfide, especially in low-dimensional form, is a subject of lively research into its electronic and optical properties. The field of twodimensional materials such as graphene and its analogues has been growing very rapidly. This class of materials also includes rhenium disulfide and rhenium diselenide, which belong to the transition-metal dichalcogenide family. Due to their reduced crystal symmetry, they exhibit distinct electrical and optical characteristics along certain in-plane crystal directions.

The group-VI members such as molybdenum and tungsten are the most typical ones, but group-VII rhenium disulfide has been attracting most attention of late because of its unusual structural, electro-optical and chemical properties; especially an indirect-to-direct band-gap transition which occurs when thinned down from bulk to monolayer. The group-VI transition-metal dichalcogenides have a 1H, 2H, 3R or 1T structure, whereas ReS₂ has a distorted 1T structure which imparts an in-plane anisotropy to its physical properties. Few other materials (black phosphorus, ReSe₂, TiS₃, ZrS₃) exhibit such an in-plane structural anisotropy. This makes ReS₂ unique among the transition-metal chalcogenides. Atomically thin rhenium disulphide is characterized by weak interlayer coupling and a distorted 1T structure, which leads to the anisotropy in optical and electrical properties. It also possesses structural and vibrational anisotropy, layer-independent electrical and optical properties and metal-free magnetism. In these respects, it differs from group-VI transition-metal dichalcogenides such as MoS₂, MoSe₂, WS₂ and WSe₂. It is already being used in solid-state electronics, catalysis, energy storage and energy-harvesting applications.

